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Formulation and Stabilization of Biotherapeutics
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Formulation of Biotherapeutics: Introduction

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M.I.T.

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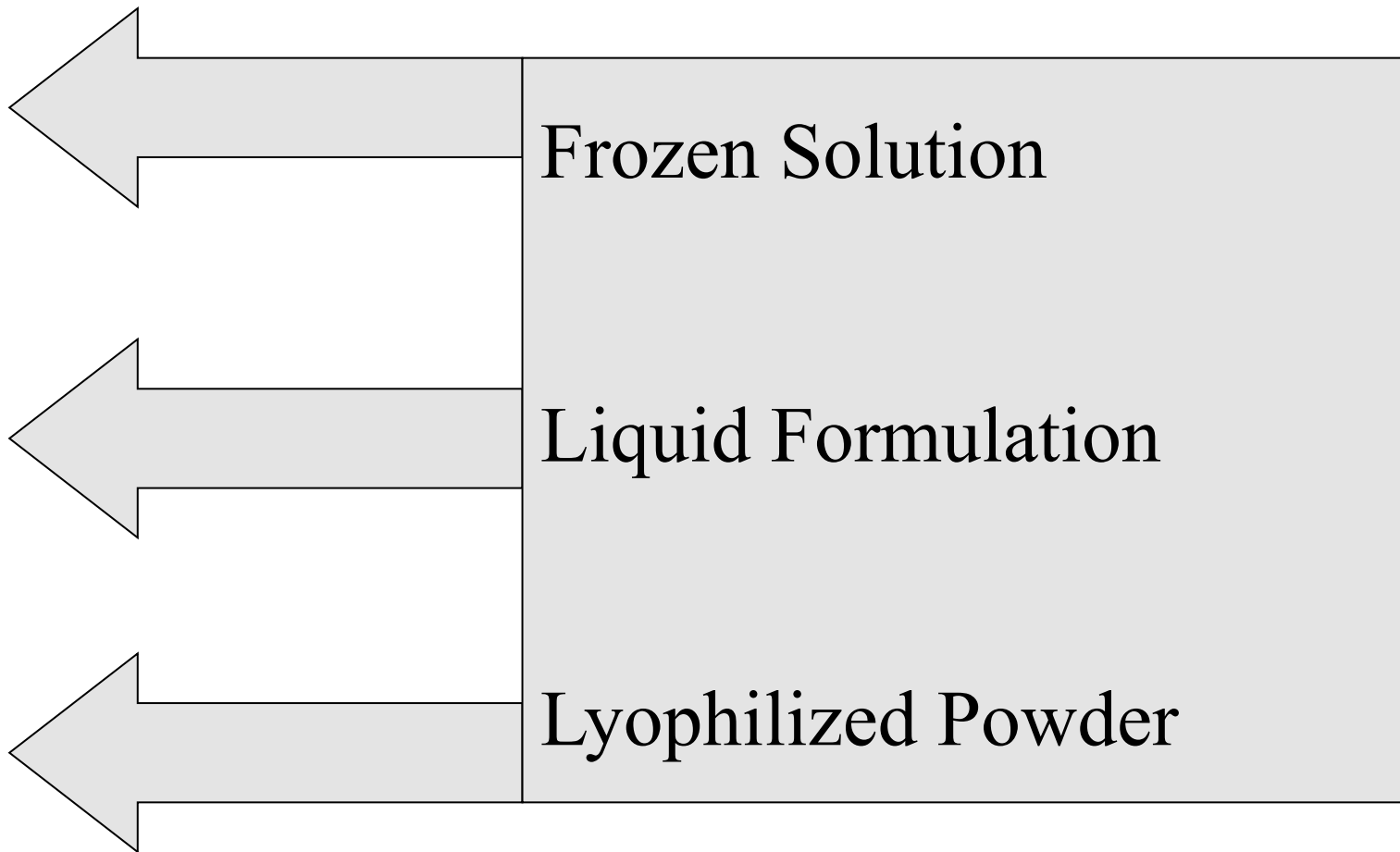
- Appreciate the importance of formulation development in bringing a biopharmaceutical product to market
- Identify the primary aspects of stabilization of proteins against physical instability (e.g., aggregation)
- Learn about the primary mechanisms of chemical degradation
- Understand the mechanisms of aggregation and the impact on kinetic profiles

- Examine the analytical methods employed to detect and quantify the levels of soluble aggregates and subvisible particles in protein products
- Use these mechanistic insights to develop liquid, frozen, and lyophilized dosage forms of proteins in a rational and defensible manner
- Learn about some important nuances for stabilization of peptides, vaccines, ADCs, and PEGylated proteins

- **To ensure optimal product ‘performance’ by the appropriate choice of (i) dosage form, (ii) excipients, (iii) process and (iv) packaging**
- The latter are done in collaboration with other groups
- What is ‘performance’ ?
 - Quality as measured by stability, solubility, potency, safety, etc.

- Develop a formulation strategy
- Obtain and meet the target profile
- Select dosage form
- Make excipient choices
- Assay development
- Employ accelerated stress testing/Select conditions
- IP constraints/goals
- Determine compatibility of formulation and container
- Ensure process and formulation are compatible

Three Primary Dosage Forms



- ◆ Simple (that's why good formulations look easy)
- ◆ Manufacturing-friendly
- ◆ Safe (use appropriate excipients)
- ◆ Meets shelf-life objectives
- ◆ Satisfies target profile
- ◆ Meets time lines
- ◆ May involve 'informed compromise'
- ◆ Must be rational/defensible

Two Types of Protein Instability



CHEMICAL

- Deamidation
- Asp-isoAsp Interconversion (Asp Isomerization)
- Racemization/Epimerization
- Proteolysis
- Trp Hydrolysis
- Hinge Region Hydrolysis
- Beta-elimination
- Oxidation

PHYSICAL

- Denaturation
 - Aggregation
 - Precipitation
 - Surface Adsorption
-
- Disulfide Exchange
 - DKP Formation
 - Condensation Reactions
 - pGlu Formation

Differences of Proteins from Small Molecules

- 1 Proteins are multi-functional
- 2 Most proteins adopt a globular structure that is essential for activity
- 3 Molecular weight differences
- 4 Numerous chiral centers in proteins
- 5 Unique pH response for each protein
- 6 Proteins are immunogenic

- Increased desirability for marketing
- Chemical stability is harder to control
- Excipients may include excluded solutes, surfactants, buffers, chemical stabilizers, others
- Damage due to agitation is a potential issue for transport and handling
- Need to evaluate F/T stability
- Higher protein concentration formulations raise additional issues

➤ Conformational Stability

Is the 3-D structure maintained?

➤ Colloidal Stability

How strong are protein-protein interactions?

➤ Interfacial Stability

Does exposure to interfaces cause damage?

- Physical stability is better (less aggregation)
- Chemical stability is easier to control (no water)
- Excipients may include lyo protectants, buffers, bulking agents, sometimes others
- Room temperature stable product is possible
- No worry about agitation-induced damage during shipping
- Higher protein concentration formulations raise additional issues

- All formulation development is assay limited
- Structural tools are essential for proper development of protein formulations (but have limitations)
- HPLC is still the centerpiece of stability testing (RP, SEC, IEX)
- Different analytical methods are needed for liquid development vs. dried formulations
- ***Critical issue is knowing which method is needed for each purpose or need***